# $N, N, N^{\prime}, N^{\prime}$-Tetramethylbenzidine-Chloranil (1:1) 

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#### Abstract

Triclinic, $P \overline{1}, a=8.427$ (5), $b=8.938$ (3), $c=$ 7.932 (5) $\AA$, $\alpha=93.36$ (4), $\beta=109.60$ (4), $\gamma=72.22$ (4) ${ }^{\circ}$, $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{2} . \mathrm{C}_{6} \mathrm{Cl}_{4} \mathrm{O}_{2}, Z=1, D_{x}=1 \cdot 51, D_{m}=1.51 \mathrm{~g} \mathrm{~cm}^{-3}$. Crystals were grown in cyclohexane solution. The TMBD molecule in this complex has a planar structure. TMBD and chloranil are stacked alternately along the $c$ axis, making their molecular planes almost parallel to each other. The mean separation of the molecular plane is $3 \cdot 29 \AA$.


Introduction. The ( $1: 1$ ) complex gives purple-black, needle-shaped crystals. The crystal chosen was less than 0.5 mm across, $\mu=5.94 \mathrm{~cm}^{-1}$ (Mo $K \alpha$ ). The cell

Table 1. Atomic coordinates and thermal parameters
(a) Atomic coordinates in fiactions of cell edges and their standard deviations $\left(10^{-3} \AA\right)$.

|  |  | $C$ | $Z$ |
| :---: | :---: | :---: | :---: |
| Chloranil | $X$ | $Y$ | $Z$ |
| Cl(1) | $0.3379(3)$ | $0.1069(3)$ | $0.0887(4)$ |
| $\mathrm{Cl}(2)$ | $-0.0109(3)$ | $-0.3122(3)$ | $-0.1988(3)$ |
| $\mathrm{C}(1)$ | $0.1515(10)$ | $0.0522(10)$ | $0.0458(11)$ |
| $\mathrm{C}(2)$ | $-0.0141(11)$ | $-0.1396(10)$ | $-0.0915(11)$ |
| $\mathrm{C}(3)$ | $0.1485(10)$ | $-0.0944(10)$ | $-0.0478(11)$ |
| $\mathrm{O}(1)$ | $0.2795(7)$ | $-0.1765(8)$ | $-0.0813(8)$ |
| TMBD |  |  |  |
| $\mathrm{C}(4)$ | $0.0653(10)$ | $-0.0562(10)$ | $0.4696(10)$ |
| $\mathrm{C}(5)$ | $0.2338(10)$ | $-0.0447(11)$ | $0.5036(11)$ |
| $\mathrm{C}(6)$ | $0.3621(12)$ | $-0.1502(11)$ | $0.4473(12)$ |
| $\mathrm{C}(7)$ | $0.3306(11)$ | $-0.2831(10)$ | $0.3530(11)$ |
| $\mathrm{C}(8)$ | $0.1631(10)$ | $-0.2998(10)$ | $0.3235(11)$ |
| $\mathrm{C}(9)$ | $0.0382(9)$ | $-0.1903(10)$ | $0.3798(11)$ |
| $\mathrm{C}(10)$ | $0.6245(13)$ | $-0.3631(12)$ | $0.3162(13)$ |
| $\mathrm{C}(11)$ | $0.4211(13)$ | $-0.5270(11)$ | $0.2064(14)$ |
| $\mathrm{N}(1)$ | $0.4575(9)$ | $-0.3899(8)$ | $0.2978(9)$ |

constants were determined by least-squares calculation based on 13 diffraction lines measured with a Geigerflex diffractometer. Quartz powder was used as the interial standard. Since no systematic absences were observed, the space group is either $P \overline{1}$ or $P 1$. The intensities were measured on a Rigaku-Denki computercontrolled four-circle automatic single-crystal diffractometer, with Mo $K x$ radiation and a crystal monochromator. The intensities were corrected for background, Lorentz and polarization effects. All reflexions with $2 \theta \leq 50$ were measured. 973 independent reflexions for which $\left|F_{o}\right| \leq 3 \sigma\left(F_{o}\right)$ were obtained. The space group $P \overline{1}$ was tentatively assumed and this was later verified by structure-factor calculations. The


Fig. 1. Projection of the molecular arrangement onto the (100) plane.

Table 1 (cont.)
(b) Thermal parameters. The $\beta$ 's are those used in the expression $\exp \left[-\left(h^{2} \beta_{11}+k^{2} \beta_{22}+l^{2} \beta_{33}+h k \beta_{12}+h l \beta_{13}+k l \beta_{23}\right)\right]$.

| Chloranil | $\beta_{11}$ | $\beta_{22}$ | $\beta_{33}$ | $\beta_{12}$ | $\beta_{13}$ | $\beta_{23}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Cl}(1)$ | 0.01705 | 0.02057 | 0.03622 | -0.01839 | 0.02600 | -0.01736 |
| $\mathrm{Cl}(2)$ | 0.02173 | 0.01396 | 0.03244 | -0.01348 | 0.02221 | -0.02190 |
| $\mathrm{C}(1)$ | 0.01375 | 0.01369 | 0.02175 | -0.01368 | 0.01349 | -0.00721 |
| $\mathrm{C}(2)$ | 0.01736 | 0.00876 | 0.02411 | -0.00593 | 0.02112 | -0.01229 |
| $\mathrm{C}(3)$ | 0.01143 | 0.01219 | 0.02585 | -0.00149 | 0.01685 | -0.01150 |
| $\mathrm{O}(1)$ | 0.01319 | 0.02103 | 0.03094 | -0.00386 | 0.02362 | -0.02201 |
| TMBD |  |  |  |  |  |  |
| $\mathrm{C}(4)$ | 0.01452 | 0.01258 | 0.01561 | -0.00587 | 0.01521 | -0.00284 |
| $\mathrm{C}(5)$ | 0.01063 | 0.01420 | 0.02289 | -0.01039 | 0.01030 | -0.01487 |
| $\mathrm{C}(6)$ | 0.01801 | 0.01551 | 0.02940 | -0.01415 | 0.02381 | -0.01491 |
| $\mathrm{C}(7)$ | 0.01921 | 0.01148 | 0.02031 | -0.00942 | 0.02537 | -0.00693 |
| $\mathrm{C}(8)$ | 0.01224 | 0.00917 | 0.02765 | -0.00917 | 0.01352 | -0.00884 |
| $\mathrm{C}(9)$ | 0.00624 | 0.1431 | 0.02883 | -0.00572 | 0.01646 | -0.01290 |
| $\mathrm{C}(10)$ | 0.02031 | 0.02175 | 0.02878 | -0.01197 | 0.02693 | -0.01776 |
| $\mathrm{C}(11)$ | 0.02423 | 0.01402 | 0.03830 | -0.00737 | 0.03840 | -0.02520 |
| $\mathrm{~N}(1)$ | 0.01445 | 0.01341 | 0.03024 | -0.00496 | 0.02832 | -0.01689 |

structure was solved by the Patterson method and was refined by block-diagonal least squares to give a conventional $R$ index of 0.093 . The weighting scheme was as follows: $w=1 /\left[\sigma^{2}\left(F_{o}\right)+\left(c F_{o}\right)^{2}\right]^{1 / 2}$, where $\sigma$ is the standard deviation estimated from the counting statistics and $c$ is the parameter which accounts for other types of error (Grant, Killean \& Lawrence, 1969). After a few least-squares calculations, the value of $c$ was determined as 0.075 . At the same time least-squares calculations with space group $P 1$ were carried out. The $R$ value became $0 \cdot 11$. The atomic coordinates did not


Fig. 2. Bond lengths ( $\AA$ ), bond angles ( ${ }^{\circ}$ ) and their e.s.d.'s, and relative orientation of TMBD with respect to chloranil found in the crystal.
change significantly, but the standard deviations in the atomic parameters increased. The usual ( $F_{o}-F_{c}$ ) synthesis with space group $P \overline{1}$, including hydrogen atoms, did not show any anomalous peaks. The space group $P \overline{1}$ was therefore adopted.*

Final parameters with their standard deviations are given in Table 1. Although the values of the $\mathrm{C}-\mathrm{H}$ bond lengths are in a reasonable range, $0 \cdot 84-1 \cdot 18 \AA$ (mean value $0.97 \AA$ ), we have omitted the hydrogen ator: ; from Table 1 because of the considerably large standard deviations in their positional parameters. Fig. 1 shows the arrangement of the molecules in the crystal. The equations of the molecular planes of TMBD and chloranil were calculated by the least-squares method. The equations are

$$
-0 \cdot 1009 X-0.4905 Y+0 \cdot 8302 Z=3.293
$$

for TMBD (without the methyl groups) and

$$
-0 \cdot 0634 X-0.4603 Y+0 \cdot 8318 Z=0 \cdot 0
$$

for chloranil, where $X, Y$ and $Z$ are coordinates in $\AA$ with respect to the crystal axes $a, b$ and $c$. The planes of TMBD and chloranil make an angle of about $2 \cdot 6^{\circ}$. The relative orientation of TMBD and chloranil is shown in Fig. 2, with the bond lengths and angles.

Discussion. We have already reported the structure of the TMBD-chloranil (2:1) complex (Yakushi, Ikemoto \& Kuroda, 1971). We determined the crystal structure of the ( $1: 1$ ) complex in order to compare, in particular, the relative orientations of the donor and acceptor molecules in the $(2: 1)$ and ( $1: 1$ ) complexes. In the polarized absorption spectra of this complex, it has been found that the direction of the transition moment of the second charge-transfer band is nearly perpendicular to that of the first. We have carried out a theoretical investigation of the electronic transitions of the TMBD-chloranil (2:1) and (1:1) complexes, and have found that the difference in the polarizations of the second charge-transfer bands can be understood on the basis of these molecular arrangements.

The HITAC 5020E computer at the Computer Centre, University of Tokyo, was used with the UNICS (1967) programs.

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[^0]:    * A table of obser ved and calculated structure factors has been deposited with the National Lending Library, England, as Supplementary Publication No. SUP 30190 ( 6 pp.). Copies may be obtained from the Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CHI 1 NZ, England.


    ## References

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